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Efficient Passivation and Low Resistivity for p⁺-Si/TiO₂ Contact by Atomic Layer Deposition

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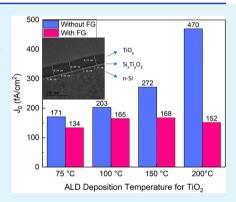
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ABSTRACT: The monolithic, two-terminal (2-T) perovskite/silicon tandem solar cell is a promising candidate to increase the power conversion efficiency beyond the theoretical limit of 29.4% for crystalline silicon solar cells. To achieve a high-efficiency 2-T tandem, it is critical to have an interface that can connect the bottom and top subcells together so that both efficient passivation and good electrical contact are achieved. The majority of works done to date in this area, applied an intermediate layer as the recombination layer between perovskite and silicon, which incurs higher manufacturing costs and an additional processing step. Here we demonstrate a unique and straightforward interlayer-free approach to passivating highly boron-doped low-resistivity n-Si using a thin layer of TiO₂ fabricated by atomic layer deposition (ALD) and a suitable pretreatment of the silicon surface. The passivation of this film is found to be superior to that of thermally grown SiO₂ formed at high temperatures over 700 °C. The TiO_X layer leads to a sufficiently low contact resistance of 0.45 Ω ·cm² and high-



quality passivation with a recombination current density (J_0) of 152 fA/cm². The structure is applicable to both perovskite/Si tandems and single-junction Si solar cells.

KEYWORDS: surface passivation, boron-doped silicon, titanium oxide, forming gas (FG), lifetime

1. INTRODUCTION

Two of the most crucial tasks that the photovoltaic industry is currently focusing on are increasing the module power conversion efficiency (PCE) and decreasing the production costs in order to diminish the cost/watt. One of the main factors to enhance solar cell performance is improving surface passivation, which can be achieved by depletion of minority carriers from the surface and/or reducing density of interface states.^{2,3} Two major techniques have been developed to passivate the surfaces of silicon (Si) cells: chemical passivation⁴ and field effect (or charge-induced) passivation. In the former process, the film adheres to dangling bonds of Si and makes them electrically inactive, while the latter involves repelling either holes or electrons by a large net charge in the passivation layer. Excellent passivation can be achieved through a combination of both the aforementioned methods.⁶ Apart from surface passivation, a low contact resistance is essential for achieving high efficiencies. There are two types of contact in current cell structures in industry, of which one is localized contact⁷ and the other one is full-area carrier selective contact.^{8,9} Localized contact structures can require complex and expensive patterning processes such as photolithography, which are not only very costly but also can introduce new contamination on the surface as a result of photoresist residue as well. 10 A carrier selective contact (CSC) is a contact that

extracts just one type of charge carrier (hole or electron), so that majority carriers can be transported efficiently and minority carrier recombination can be reduced. A CSC layer can provide superb surface passivation leading to an improvement in open-circuit voltage $(V_{\rm oc})$ and fill factor (FF). Thus, by eliminating the need for localized doping and opening the contact, a simplified and flexible fabrication process can be achieved.

Having a passivation layer that can also provide good contact is particularly beneficial for two-terminal (2-T) tandem solar cells, where a low band gap bottom cell (e.g., silicon) is serially connected to a high band gap top cell (e.g., perovskite). Generally, the 2-T configuration requires an interlayer between the bottom and top subcells to facilitate the current flow of photogenerated carriers. The aforementioned interlayer should have both low electrical and optical losses with easy fabrication processing. Recently, a type of unconventional

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perovskite/Si tandem structure that bypasses the use of an interlayer was developed, 12,13 with the advantage of not only reducing the fabrication steps but also eliminating the optical losses associated with highly doped recombination layers. Tandem cell efficiencies > 24% were achieved by applying this new concept to heterojunction Si subcells, while efficiencies approaching 23% were demonstrated with a Si homojunction subcell. Tandem cells based on homojunction Si technology are potentially most relevant to the PV industry given their dominance in commercial products.¹⁴ This interlayer-free design relies on a surprisingly low ohmic contact resistance between p+-Si/TiO2, which enables in situ recombination on the atomic layer between p+-Si and TiO2. In our previous results, the processing conditions required to achieve a lowresistivity p+-Si/TiO2 contact provided only limited surface passivation of the p+-Si. This correspondingly induced a high recombination rate in the front emitter of the bottom Si solar cell, thus limiting its contribution to the tandem efficiency. As discussed above, high-efficiency solar cells require interfaces that provide both low contact resistance and excellent surface passivation, which is the focus of this work.

Passivation of a p+-emitter on an n-type Si wafer diffused with boron tribromide (BBr₃) is challenging 15 owing to high boron solubility and high concentration of built-in positive charges, 16 and development of a low-cost technique to passivate p⁺ emitters is still demanded by industry.¹ Negatively charged dielectric layers are normally favored options to passivate these surfaces, ¹⁶ as they can provide an electric field that repels the minority electrons from the interface and thus suppress recombination.⁶ Various methods have been used to passivate boron-diffused surfaces, such as hydrogenated amorphous silicon (a-Si:H), ¹⁸ thermally grown ¹⁹ or chemically grown SiO₂, ¹⁷ Al₂O₃, ¹⁶ TiO₂, ^{10,20} hafnium oxide, ²¹ tantalum oxide, ²² and gallium oxide²³ deposited by atomic layer deposition (ALD), plasma-enhanced chemical vapor deposition (PECVD) of SiO_x/SiN_y stacks⁶ or AlO_y²⁴ sputtered aluminum nitride, 25 and stacks of Al₂O₃/SiN_x. Although because of the insulating nature of some of the abovementioed materials, it makes it necessary to have the contact opening in the structure. Due to its small conduction band offset with silicon,²⁷ insignificant extinction coefficient, negligible parasitic light absorption, and very low surface recombination velocities accompanied by low contact resistances, TiO₂ has drawn significant attention as an electron selective contact²⁸ for Si solar cells, which has the potential to lift crystalline Si solar cell efficiency to over 26%. Other attractive features of TiO2 as a CSC are low-cost and nontoxic processing options via atomic layer deposition, thermal evaporation, or spin-coating.²⁸ Yang et al. demonstrated surface passivation using an ultrathin SiO₂/TiO₂ stack for the rear side of undiffused n-type silicon solar cell. 10 They reported low contact resistance of 0.02 Ω·cm² with surface recombination velocity of 16 cm/s for 2.5 nm titanium oxide deposited by ALD at 75 °C. Cui et al. studied surface passivation of boron diffused 100 Ω·cm n-type FZ silicon wafers (with sheet resistance of 120 Ω/sq) using ALD TiO₂ deposited at 75 °C.²⁰ They found that increasing the thickness of TiO₂ to around 15 nm improved the passivation to a J_0 of 19 fA/cm², and this value remained almost constant for thicker films. In contrast to these two previous works, and on the basis of our previous work¹² which showed that lowering the boron doping will largely increase the contact resistivity, here we focus on passivation of heavily boron doped silicon to improve

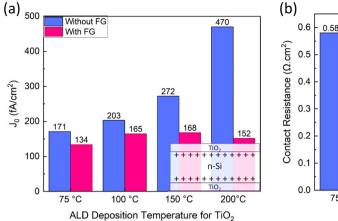
the electrical contact, which is more challenging as mentioned above. As we intend to improve the front side passivation of silicon solar cells for future application in 2-T tandem cells, the importance of surface passivation outweighs the contact resistance, which led us to choose thicker TiO_x films. In addition, for tandem solar cells, a low-temperature process ($\leq 200~^{\circ}\mathrm{C}$ for heterojunction and $\leq 400~^{\circ}\mathrm{C}$ for homojunction silicon subcell⁷) is desirable to minimize damage to the bottom silicon cell, ¹³ which motivated us toward finding a fabrication method compatible with relatively low temperatures.

In this context, we focus on improving the interface of p⁺-Si/ TiO₂ by controlling the fabrication conditions as well as implementing pretreatment and post-treatment, with the aim of achieving effective surface passivation of heavily boron doped silicon and low contact resistance simultaneously. There are various methods for TiO₂ deposition such as evaporation,²⁵ spray hydrolysis,³⁰ sol-gel,³¹ sputtering,³² atmospheric pressure chemical vapor deposition (APCVD)³³ and, recently, ALD. 20,34 Here, we selected ALD to deposit the TiO₂ film, as it can offer high surface uniformity and controlled growth rate at the atomic scale by exploiting sequential, self-limiting reactions between gaseous precursors and the substrate. Consequently, outstanding conformal and uniform films can be obtained on a range of surfaces with precise thickness in the range of angstroms. 35,36 We developed a novel, cost-effective method to improve the passivation of TiO2 films on heavily boron doped surfaces of n-type crystalline Si solar cells using a combination of forming gas (FG) preheat treatment of Si substrate followed by ALD deposition of ultrathin TiO₂ at 200 °C. An important advantage of this approach is that it can be applied to the interface layer between p⁺-Si and TiO₂ in a perovskite-silicon tandem without the need for an additional recombination layer. The FG annealing of the boron-doped silicon substrate, prior to subsequent depositions, is a straightforward, relatively low temperature process which improved the ligand exchange and bond formation between silicon and TiO2. Our results show that FG annealing can reduce recombination by a factor of 3 at a TiO₂ deposition temperature of 200 °C, while also providing acceptably low contact resistance. We note that the work here is not only appropriate for use in 2-T siliconperovskite tandem solar cells but also for single-junction silicon solar cells.

2. EXPERIMENTAL SECTION

Surface passivation was investigated on 5 Ω ·cm n-type float-zone (FZ) crystalline silicon (c-Si) with a thickness of 400 μ m and 2.5 Ω ·cm p-type FZ c-Si wafers with a thickness of 300 μ m, with and without boron (p⁺) and phosphorus (n⁺) diffusion. Samples for contact resistance investigation were fabricated on boron-diffused p-type Czochralski (Cz) c-Si (1 Ω cm) wafers with a thickness of 250 μ m.

Initially all wafers were etched in 25 wt % tetramethylammonium hydroxide (TMAH) at approximately 80 °C for 5 min to remove any saw damage followed by a standard RCA (Radio Corporation of America) cleaning process 37 and a 1% (w/v) HF (hydrofluoric acid) solution dip. Boron and phosphorus diffusion processes were carried out in quartz tube furnaces using liquid BBr $_3$ and POCl $_3$ precursors, at 980 and 900 °C, respectively. Four-point-probe measurements gave sheet resistances of 23 $\Omega/{\rm sq}$ for boron-diffused and 13.6 $\Omega/{\rm sq}$ for phosphorus-diffused samples, corresponding to surface doping concentrations $7\times10^{19}~{\rm cm}^{-3}$ and $2\times10^{20}~{\rm cm}^{-3}$, respectively. The resultant borosilicate and phosphosilicate glasses layers were then removed using an HF dip process. Another RCA cleaning and HF dip were applied on the wafers prior to preheat treatment and ALD-TiO $_2$ deposition.



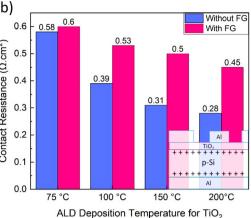


Figure 1. Effect of different ALD deposition temperatures (from 75 to 200 °C) and the FG annealing pretreatment on (a) surface passivation and (b) contact resistance with the inset sketches of the tested structures.

Various preheat treatment procedures (under high or low temperature) were conducted prior to ${\rm TiO_2}$ deposition, and their passivation effects were compared. Different types of atmosphere were utilized comprising annealing in FG, N₂, and O₂ at 400 °C for 30 min for the low-temperature procedure and annealing in O₂ at 700 °C for 3 min for the high-temperature process. For some samples we directly deposited ${\rm TiO_2}$ without any preheat treatment.

TiO₂ deposition by a thermal ALD system (TFS200, BENEQ) was carried out using titanium tetrachloride (TiCl₄) as the titanium precursor, deionized water as the oxidant agent, and pure nitrogen (N_2) as the purge gas. The reactor temperature was initially set to be 230 °C, but by decreasing the temperature to 200 °C, the minority carrier lifetime improved from around 900 to 1142 µs at a carrier density of 1.0×10^{15} cm⁻³. At this stage 200 °C was chosen as the main deposition temperature for further study; however, lower temperatures (including 75, 100, and 150 °C) were examined as well. The deposition process consisted of 200 sccm (standard cubic centimeters per minute) N₂(g) flow, with a 0.05 s pulse of TiCl₄ followed by a 0.05 s pulse of H₂O per cycle, and a 0.75 s purge under a constant flow (300 sccm) of N₂(g) between each precursor pulse. For a deposition temperature of 75 $^{\circ}$ C, a deposition rate ~ 0.76 Å/ cycle was measured by spectroscopic ellipsometry (M2000, J .A. Woolam).

The surface passivation behavior was studied using a Sinton Instruments WCT-120 photoconductance tool in the transient mode which is not dependent on the optical properties of the sample.³⁸ The recombination current density was extracted at $\Delta n = 3.0 \times 10^{15}$ cm⁻³, considering the radiative recombination model of Nguyen et al.³⁹ at 300 K. The electrical contact resistance was evaluated by the Cox and Strack method, 40 in which TiO₂ was deposited on the front side of a double-sided boron-diffused wafer. Subsequently, an array of Al circles with various diameters (0.1, 0.15, 0.2, 0.25, 0.3, 0.35, 0.4, 0.5, 0.6, 0.7, and 0.8 cm) were evaporated on the front side via a shadow mask. The back side of the wafer was coated completely with an Al film. Current-voltage (I-V) measurements at room temperature were performed, which utilized a spreading resistance model to fit the resistance versus diameter data. This method calculates the contact resistance that includes the contact resistance of the metal oxide/Al, the bulk oxide, and the p+ Si/oxide.

High-resolution transmission electron microscopy (HRTEM) was used to observe the existence of the interlayer between silicon and TiO_2 , as well as measuring the thickness of these layers accurately. Cross-sectional TEM samples were prepared using standard techniques including grounding, dimpling, and ion milling. High-resolution TEM imaging was performed using a FEI Titan 80-300 microscope with an image-side C_s corrector. Additionally, XPS and UPS measurements were individually conducted to determine the elemental composition and electronic structure of different samples. XPS measurements were performed by recording the electrons

emitted due to X-ray radiation from the core level of the element with a pass energy of 10 eV in an ultrahigh-vacuum chamber, which provides information on the chemical composition and valence electron states of the samples. The instrument is described in detail in ref 41. Calibration of the XP spectra was not applied because no charging was observed during the measurement and the C–C bond of ${\rm TiO_2}$ samples was found close to 285.0 eV, which is the energy usually applied for calibration of XP spectra. Not applying the calibration procedure allows for identifying chemical changes of C. ⁴² An error bar for the peak fitting of XPS is normally under \pm 0.2 eV, depending on the counts and fitting.

UPS was conducted for acquiring information on work function (WF) and valence electron states. The samples were transferred to the measurement chamber, where a short heating process under vacuum conditions at 250 °C for sintered samples and 150 °C for unsintered samples was applied prior to the measurement, to remove the potential moisture and minimize the carbon contamination. The error bar for the electronic properties from UPS is ± 0.1 eV.

Low-temperature PL spectroscopy was performed with a diode-pumped solid-state (DPSS) continuous-wave 532 nm laser to confirm the surface passivation results. The emitted PL signal was captured with an InGaAs array detector. The sample temperature was controlled with a liquid-nitrogen-cooled THMS600 Linkam stage. The spectral response of the entire system was determined with a calibrated halogen—tungsten light source.

Grazing incidence X-ray diffraction (GIXRD) analyses were conducted on thin films using a PANalytical X'Pert Pro system, with a tube voltage of 30 kV and a tube current of 10 mA to generate copper $K\alpha$ X-rays.

3. RESULTS AND DISCUSSION

3.1. Surface Passivation and Contact Resistance. The passivation effect can be studied by investigating the effective lifetime of a Si wafer which determines the recombination rate, thus significantly affecting the voltage of a corresponding solar cell, while the contact resistance affects mostly the fill factor of the device. 10 In this work, we first undertook a comparative study to examine the effect of ALD deposition temperature on the surface passivation and contact resistance of different p⁺-Si/TiO₂ samples. We note that the samples here for the lifetime tests use a symmetrical structure of TiO₂/p⁺/n/p⁺/ TiO2, allowing injection-dependent lifetime measurement with straightforward calculation of the implied voltage of the cell, 10 while the samples for the p⁺-Si/TiO₂ contact test are based on a p⁺-Si/p-Si/p⁺-Si structure. By depositing a 10 nm thick TiO₂ layer at an ALD deposition temperature of 75 °C, a recombination current density J_0 of 171 fA·cm⁻² can be

reached. By increasing the ALD deposition temperature from 75 to 200 °C, J_0 gradually increased to 470 fA·cm⁻² as shown in Figure 1a. Such a trend is similar to the work presented by Cui et al., 20 where undiffused wafers were utilized. On the basis of the results reported by Aarik et al., 43,44 an ALD deposition temperature up to 100 °C results in pure amorphous TiO₂ 45 By increasing the deposition temperature from 150 °C some crystal grains appear in the structure, and by further raising the temperature, the structure of TiO2 film becomes more crystalline. Because the anatase phase is growing within the amorphous phase, hillocks on the film surface are formed by the crystallites which roughen the surface and also increase the surface area. 44,46 All of the aforementioned features in films deposited at temperatures above 150 °C increase the recombination rate by providing more recombination sites which impact the passivation quality. Although surface passivation is deteriorated at higher temperature, the contact resistance is reduced by 300 m Ω ·cm², from 580 m Ω ·cm² at 75 °C to 280 m Ω ·cm² at 200 °C (Figure 1b), which favors charge transfer at the interface. This improvement in contact resistance can be attributed to defect-mediated conductivity, where the formation of crystallites introduces some midgap states and grain boundaries so that these sites act as generation-recombination centers and facilitate the charge transfer. 12 Such a trade-off makes it hard to achieve excellent lifetime and low contact resistance simultaneously, a situation that we aimed to address by pretreatment.

Hydrogen has been widely reported to be beneficial for improving the surface passivation of Si, as it can diffuse to recombination centers in the form of atomic hydrogen and deactivate them. 10,47 A widely established passivation approach for Si is by depositing passivation layers such as SiO₂,² Al₂O₃, ⁴⁸ followed by forming gas (H₂/Ar) annealing. However, in the present work, introducing an extra passivation layer between p+-Si and TiO₂ using the above-mentioned materials is not feasible, as it would dramatically increase the contact resistance due to their insulating nature, which necessitates having metal contact openings. Instead, in this work we introduced a novel FG pretreatment for the Si wafer immediately before the ALD-TiO2 deposition and found that this simple pretreatment enhanced the surface passivation substantially. We hypothesize that pretreatment by FG forms hydroxyl groups on the Si surface which facilitate a ligand exchange process later during the ALD deposition of TiO₂, resulting in better interaction between the Si substrate and TiO₂. The initial surface passivation test on the boron-doped silicon wafer without TiO2 deposition revealed that FG annealing slightly improved J_0 from the initial value of 961 fA·cm⁻² for p⁺-Si to 856 fA·cm⁻² for p⁺-Si/FG, which proves that FGA is unable to passivate the silicon surface by its own, but instead it affects the following ALD deposition and enhances passivation quality. The J_0 reduction resulting from FG pretreatment was ~40 fA·cm⁻² for ALD-TiO₂ deposited at 75 and 100 °C and \sim 100 fA·cm⁻², for ALD-TiO₂ deposited at 150 °C, as shown in Figure 1a) The improvement is most significant for the 200 °C ALD-TiO2, with an almost 3-fold reduction of J_0 from 470 to 152 fA·cm⁻² (Figure 1a). Moreover the effect of TiO2 thickness on surface passivation for a film deposited at 200 $^{\circ}\text{C}$ was examined (result is presented in Figure S1 in the Supporting Information), which revealed the 10 nm as the optimum thickness. The resultant improvement from FGA is stable with time; not only did the lifetime not drop after 9 weeks storage in ambient condition but also it

improved moderately (refer to Figure S2 in the Supporting Information). The low-temperature steady-state PL spectra demonstrated results consistent with the trends in I_0 values, with the maximum peak intensities from both the Si substrate and the p⁺ layer observed for the p⁺-Si/FG/TiO₂ sample, and intensities more than six times lower for p+Si, p+Si/FG, and p⁺-Si/TiO₂ samples (refer to Figure S3 in the Supporting Information and ref49 for the interretation of each peak). This indicates significantly lower trap assisted recombination for both the Si substrate and the p+ layer in the FG pretreated samples, which can be attributed to the formation of a hydrophilic hydroxyl group on the silicon surface that affects subsequent TiO2 deposition. Meanwhile, we found such a treatment only moderately increased the contact resistance between p⁺-Si and TiO₂ as shown in Figure 1b), which makes it practicable to have full-area carrier selective contact. Moreover, the contact resistance is still superior to that which is mentioned in the previous work for the similar structure $(2 \Omega \cdot \text{cm}^2)$. It should be noted that, for perovskite/ Si tandems, the issue of series resistance is much less severe relative to silicon cells (by a factor of $\sim 5^{50}$), due to the higher voltage and lower cell current, resulting from the necessity for the subcell currents to be matched. Hence, all contact resistivities obtained with the FG pretreatment are compatible with high-efficiency perovskite/Si tandem cells, and the surface passivation improvement does not have to be traded off against severe resistance losses. In addition, the lateral conductivity is not so high to cause shunting.

Some of the best Si surface passivation approaches are based on high-quality SiO_2 fabricated by high-temperature oxidation $(700-1000~^{\circ}C).^{51}$ We therefore compared the passivation quality of our low-temperature passivation approach with these standard high-temperature passivation approaches. On the basis of the results, the passivation quality of samples with FG pretreatment is comparable and even slightly better than the samples with high-temperature oxidation in our experiments (refer to Figure S4 in the Supporting Information).

The same trend of improved passivation with FG pretreatment was observed among the samples without boron diffusion, in which FG pretreatment enhanced the passivation quality, although with relatively lower lifetime arising from the higher densities of the minority carriers in undoped samples, which leads to stronger impact of the surface defects on total recombination.¹⁵ We found that FG annealing resulted in a much smaller improvement for the phosphorus-diffused samples, which can be explained by the possibility of electron depletion resulting in an accumulation of holes on the n+ surface, resulting in a high recombination rate (refer to figure S5 in the Supporting Information). This might indicate the presence of negative charge in the TiO₂ film, ²⁰ but we have been unable to confirm this hypothesis with capacitancevoltage (CV) measurements because the high conductivity of the TiO2 results in a large current leakage.

3.2. High-Resolution Transmission Electron Microscopy Analysis. In order to examine the interface layer between the silicon substrate and TiO₂ layer as well as investigate the crystallization within the TiO₂ layer, HRTEM analysis was carried out on two different samples with and without FG pretreatment. Boron-diffused n-type samples were used with 30 min FG annealing at 400 °C for one sample, followed by TiO₂ deposition by ALD at 200 °C for both samples. TEM imaging reveals the existence of an amorphous interlayer between p⁺Si and TiO_x for both types of samples,

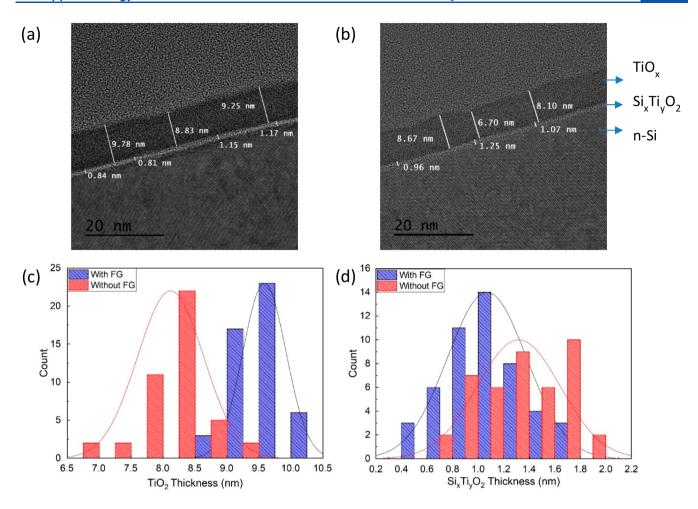


Figure 2. TEM cross-section images of (a) $p^+n-Si/FG/TiO_2$ and (b) p^+n-Si/TiO_2 , and histogram graphs of (c) TiO_2 and (d) $Si_xTi_yO_2$ thickness distributions.

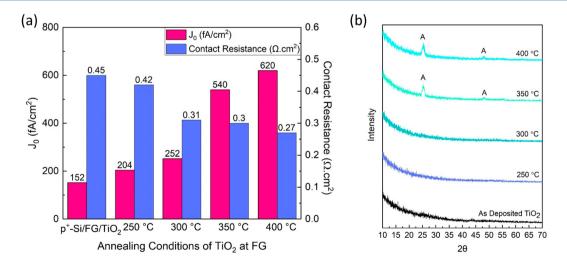


Figure 3. (a) Effect of postannealing temperature on surface passivation and contact resistance for the TiO_2 layer deposited at 200 °C and (b) the corresponding XRD pattern for TiO_2 different annealing temperatures (A = anatase).

which is a mixture of Si, Ti, and O (termed as $\mathrm{Si_xTi_yO_2}$ here) (refer to Figure S6 in Supporting Information). Careful analysis of the high-resolution TEM images indicates higher crystallinity of $\mathrm{TiO_x}$ in samples without FG in which crystal grains appear as bright features in dark field images. The density of crystallites was calculated by Gatan Digital

micrograph software. While the measured density of crystallites for the sample without FG was around $0.011~\text{nm}^{-2}$, for the sample with FG it was almost half $(0.005~\text{nm}^{-2})$ with mostly amorphous structure. This may be a reason for the superior surface passivation of samples with FG, since crystallization will introduce grain boundaries as extra recombination sites. ¹⁰

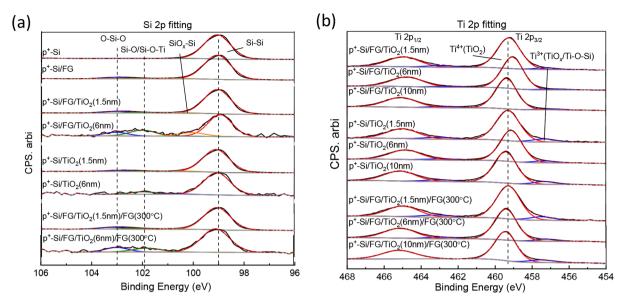


Figure 4. XPS result for (a) Si 2p and (b) Ti 2p orbitals for p+n-Si/TiO2 samples with different TiO2 thicknesses.

We measured the thickness of the intermediate layer as well as the TiO_x layer in 10 different images for each condition. As can be seen from TEM images and histograms in Figure 2, the average thickness of titania in samples with FG treatment is around 9.5 nm, slightly higher than without FG annealing (~8.2 nm). A possible interpretation of this trend is that samples without FG annealing contain a hydrogen-terminated silicon bond at the surface while samples with FG have hydroxyl-terminated silicon bond in the surface. Whereas the hydrophobic nature of the former makes it difficult to adsorb the TiCl₄ precursor, the latter with a hydrophilic nature is reactive toward TiCl₄ adsorption. This would make the ligand exchange process easier between hydroxyl-terminated silicon and the gaseous precursors, resulting in higher growth rate of TiO₂ on FG annealed substrates similar to what is reported by Methaapanon and Bent.³⁵ On the other hand, on the basis of the histograms (Figure 2d), the intermediate layer in samples without FG is a little thicker (\sim 1.3 nm compared to \sim 1.1 nm). The logical conclusion that can be drawn is that in ALD the initial cycles were utilized to prepare the surface bonds by making an interface layer, and for substrates without FG treatment more cycles were consumed to complete the ligand exchange process and make the intermediate layer prior to actual deposition of TiO₂.3

3.3. Effect of Post-treatment. Some 2-T tandem cell architectures require a further annealing step for the perovskite top cell; for those structures it will be necessary to maintain the passivation quality of the Si cell. Because the postannealing can change the crystallization of the as-prepared TiO2 film and thus change the interface properties, we examined the effect of postannealing of TiO2 layers in FG at various temperatures on phase composition, which can alter surface passivation and contact resistance. Figure 3a shows that there is a gradual decrease of contact resistance and a concomitant increase in J_0 with increasing annealing temperature. Annealing at 350 and 400 °C results in dramatic loss of the surface passivation to 540 fA/cm² and 620 fA/cm², respectively. By making a comparison between the J_0 results and XRD patterns in Figure 4b, no peaks are observed for the as-deposited TiO₂ samples as well as TiO₂ samples annealed at 250 and 300 °C, indicating low crystallization that is beyond the detection level for XRD.

Upon further increasing the annealing temperature to above 350 °C, peaks at 25 and 49° appear, representing the anatase phase in the (101) direction. This is the main reason for the dramatic decrease in surface passivation under these treatments, as it has been shown previously that the formation of crystal grains resulting from phase transformation from amorphous to anatase introduces grain boundaries which accelerate recombination. 10 In addition, annealing the sample at temperatures above 300 °C might break the Si-O-Ti bonds and form some dangling bonds, 10 which could also contribute to the reduction in surface passivation. Yang et al. deposited an ultrathin layer of ALD-TiO_x with up to 5.5 nm thickness on silicon substrates followed by annealing in FG for 30 min.²⁷ They reported that by postannealing the samples at 250 °C, the lifetime improved; however, a substantial passivation degradation was observed after FG annealing at temperatures \geq 300 °C. The reason for the difference between their work and the current one can be attributed to the greater thickness of TiO2 in our work which can accelerate the nucleation process,²⁷ and as a result no passivation enhancement is obtained even after annealing at 250 °C. For the tandem structure the effect of surface passivation is more important than the contact resistance. We can estimate how much each postannealing condition will affect the performance of a 2-T tandem solar cell as follows. We used the measured contact resistance and implied $V_{\rm oc}$ values for each p⁺-Si/TiO₂ condition in Figure 3a to estimate the tandem efficiency that would result with each structure in place of a reference structure used in our previous work. 12 Specifically, we applied the joint corrections of an adjusted V_{oc} (based on difference in iV_{oc} values between each structure and that of the reference measured at 606 mV) and adjusted series resistance (again compared to the reference contact resistivity of 2 $\Omega \cdot \text{cm}^2$) to correct the reference tandem's IV curve and thereby obtain an estimated efficiency value. Figure S7a in the Supporting Information shows this increment in (absolute) efficiency relative to the reference for each structure. On the basis of the obtained results for samples without postannealing, the absolute efficiency could improve by about 0.86% compared to the reference cell with 3.5% V_{oc} enhancement. Postannealing at 250, 300, 350, and 400 °C would increase the efficiency of the tandem cell in comparison with the reference cell by around 0.87, 0.74, 0.42, and 0.26%, respectively. Also, the J-V curve for the ${\rm TiO_2/Si}$ contact in various samples without and with postannealing of ${\rm TiO_2}$, is presented in Figure S7b) in the Supporting Information section that displays conductive but nonlinear J-V behavior.

3.4. X-ray Photoelectron Spectroscopy and Ultraviolet Photoelectron Spectroscopy Analysis. To determine elemental composition, chemical state, electronic state, and binding energy, X-ray photoelectron spectroscopy (XPS) was carried out. Due to the XPS measurement depth of ~10 nm, different samples with different thicknesses were fabricated to supply information from both the interface and the bulk of TiO₂. Moreover, in order to investigate the electronic characteristics of the samples including work function and energy level structure, ultraviolet photoelectron spectroscopy (UPS) was utilized. UPS provides information on how FG annealing can alter the work function and electronic structure in the valence electron region. Figure 4 plots the XPS result for 2p orbitals of Si and Ti for TiO₂ deposited on boron-doped n-Si samples with different thicknesses of TiO2 for an ALD deposition temperature of 200 °C; as with 10 nm of TiO2 on a Si surface, the Si was not observable from XPS and thus not plotted in Figure 4a. There are four different peaks in the Si 2p fitting diagram representing Si-Si, SiO_x-Si, Si-O/Si-O-Ti (compatible), 52,53 and O-Si-O 54,55 bonds with around 99.1, 100.2, 101.8, and 103.0 eV (\pm 0.2e V) binding energies, 35,52,56,57 respectively. It needs to be pointed out that such peaks represent a range of different oxidation states of Si, depending on its crystallite geometry with O bond incorporated. The Si binding energy increases with increasing Si oxidation state. 58,59 While the Si–Si peak is present for all samples, the other three peaks are only apparent in some of them. Although for the p⁺-Si sample the only visible peak is Si-Si, FG annealing introduces O-Si-O bond (p+Si/FG). The presence of the 103.0 eV Si peak after FG annealing indicating an O-Si-O species is compatible with the formation of ultrathin hydroxyl-enriched SiO2 as proposed by Hu and Turner.³⁶ It can be seen that, for samples with FG pretreatment, for 1.5 nm thickness of TiO₂ the peak of O-Si-O is very small but becomes more intense when the TiO₂ thickness is increased to 6 nm. Also SiO_x-Si and Si-O peaks that were absent for the thinner film can be seen for the thicker one. It should be noted that the 101.8 eV peak assigned to Si-O is also compatible with assigning it to Si-O-Ti. 54,35 Occurrence of the SiO_x-Si and Si-O peaks can be explained by the higher amount of available oxygen in the titania layer that can be used by the Si substrate to make an oxide layer in between. For samples without FG pretreatment for 1.5 nm thickness of TiO2 there is a very small peak of O-Si-O with no Si-O/Si-O-Ti (compatible) and SiO_x-Si peaks (quantified results are presented in Table S1 in the Supporting Information). However, for 6 nm TiO2 the peak intensity for Si-O/Si-O-Ti (compatible) is slightly increased. The possible reason for the lower intensity of this peak might be the hydrophobic nature of this silicon substrate which slows the formation of the Si-O-Ti bond.³⁵ For samples with annealed TiO2 (at 300 °C, FG) a similar trend can be observed in which the O-Si-O peak is smaller for the 1.5 nm sample compared with the 6 nm sample, and in addition the Si-O (identification compatible with Si-O-Ti) peak also appears in the sample containing 6 nm TiO2. The existence of these two

peaks is possibly an indication for the formation of an oxide layer at the interface at the expense of a reduction of TiO_x .

Figure 4b shows the Ti 2p fitting with four different peak positions comprising $Ti^{3+}(TiO_x/Ti-O-Si)^{60}$ Ti⁴⁺(TiO₂)⁶¹ peaks corresponding to Ti 2p_{3/2} orbital located at around 457.5 and 459.4 eV, 62 respectively. Ti⁴⁺ peaks for both orbitals are present in all conditions. For samples with FG pretreatment it can be seen that by increasing the thickness of TiO₂ from 1.5 to 6 nm, the peak intensity of Ti³⁺ decreased which can be attributed to the Ti-O-Si bond that is related to the interface and is getting hard to detect as the TiO2 is becoming thicker. The intensity of the Ti peak is thus attributed mostly to the TiO2 bulk. For samples without FG pretreatment and samples with pretreatment and post-FG annealing of TiO₂ at 300 °C, higher intensity for the Ti³⁺ peak was observed by comparison with the analogous sample with FG pretreatment (quantified results are presented in Table S2 in the Supporting Information). Ti³⁺ defects are recombination sites which deteriorate passivation and increase I_0 at the interface and can be found in the bulk as well as the interface of TiO2. On the basis of previous works, one hypothesis is that defects available on the substrate surface plus the temperature during annealing can provide enough activation energy for H2 molecules to be dissociated into H atoms and penetrate into the surface which later, after exposure to oxygen, can form a hydroxyl group on the surface. 47,63 By postannealing of TiO₂, Ti loses some of its oxygen and transforms to lower oxidation states (Ti3+) which can make some sub-band-gap states and introduces new recombination sites.³⁵ We note that the Ti⁴⁺ showed slight shift to a lower binding energy of 6 nm TiO₂ samples. This could be due to the different ALD processing batches of various thicknesses. Since the Ti⁴⁺ and Ti³⁺ peaks were still observable, such a shift would not affect the fitting of the XP spectra.⁶⁴

Figure 5 depicts the density of states (DOS) of valence electron close to the Fermi level for p^+ -Si/TiO₂(10 nm), p^+ -Si/

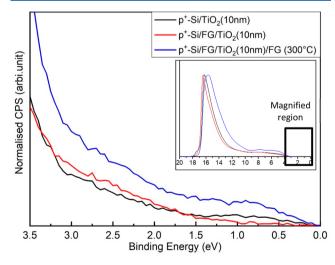


Figure 5. Density of states of valence electrons for different samples. The energy ranging from 0 to 3.5 eV is magnified.

 $FG/TiO_2(10 \text{ nm})$, and $p^+-Si/FG/TiO_2(10 \text{ nm})/FG(300 \,^{\circ}C)$. As can be seen from the graph in the range of 1.5 to 0 eV binding energy, the sample without FG pretreatment has a higher DOS of electrons at lower binding energy arising from the formation of lower electronegativity cations and surface

dipole change which give rise to the work function reduction.⁶⁵ In light of the higher density of low-oxidation-state cation (Ti³⁺) among all samples without FG pretreatment compared with their equivalent sample with FG pretreatment (Figure 5), it could be argued that O-vacancy defects available within the bulk and interface of TiO₂ are responsible for higher recombination rate in these samples. 65,66 By applying the FG annealing to the p⁺-Si/FG/TiO₂ sample (red line in Figure 5), the DOS for electrons with low binding energy decreased (evident by the decrease in Ti³⁺ defect states close to the Fermi level). Postannealing of the TiO_X layer at FG at 300 °C increased the DOS within the range of 1.5 to 0 eV binding energy. Postannealing provides some activation energy for titania reduction which introduces some unpaired electrons in the valence band of Ti with low binding energy which leads to an increase in the DOS for these electrons followed by improving the contact in this sample. The electronic characterization indicated that FG pretreatment slightly shifted conduction and valence energy level upward (refer to Figure S8 in the Supporting Information), which enlarged the band offset between the TiO2 conduction band edge and the Si valence band (Δ) from 0.4 eV in the sample without FG pretreatment to 0.45 eV in the sample with FG. Moreover, after postannealing of TiO2 in FG, the energy levels showed a downward shift that decreased the Δf actor to 0.35 eV. It is well-known that the band offset is one of the responsible factors affecting the conductivity between TiO₂ and p⁺-Si. 12 This influence on the conductivity is not compensated for by the increase in DOS close to the Fermi level due to the increase in Ti3+ defect states. Therefore, the obtained results are consistent with the contact resistance measurements in which the contact resistance was slightly better for the sample without FG and also it improved after postannealing of the sample with FG pretreatment (refer to Figure 3a). In summary XPS and UPS results demonstrate the positive effect of FG pretreatment in improving the passivation of the Si substrate by decreasing the defect density at the interface between the Si and TiO2 layer and upward shifting of the energy states of the TiO2 layer.

4. CONCLUSION

In this study we demonstrated the excellent performance of a FG pretreatment (400 °C, 30 min) in combination with ALD-TiO₂ as a passivation and potential carrier selective contact layer for highly boron doped silicon of 23 Ω /sq. With this combined treatment we achieved a J_0 of 152 fA·cm⁻² and a contact resistance of 0.45Ω·cm² for ALD-TiO₂ deposited at 200 $^{\circ}$ C. This low contact resistance eliminates further need for localized contact structures. The comparison between different ALD deposition temperatures shows a significant improvement in Si surface passivation when including the FG pretreatment irrespective of the TiO2 deposition temperature. The FG pretreatment significantly improves surface passivation without a strong increase in contact resistance. Furthermore, the passivation quality of the best FG pretreated TiO2 films exceeds that of thermally grown SiO2 (700 °C). HRTEM characterization showed higher crystallinity for samples without FG pretreatment, which is one reason for the inferior passivation quality of these samples. The results of XRD measurements show that the TiO2 samples with FG pretreatment were nearly completely amorphous. XPS results confirmed the almost stoichiometric composition of TiO2 with a lower density of defects for samples with FG

pretreatment which explains the superior passivation of these samples. We hypothesize that the pretreatment in FG forms an ultrathin interfacial silicon oxide terminated with hydroxyl groups that facilitates the immediate chemisorption of the ${\rm TiCl_4}$ precursor and thereby stoichiometric, defect-lean ${\rm TiO_2}$ deposition without delay (no incubation cycles). Also, the ${\rm H_2}$ molecules in the forming gas can dissociate into atomic hydrogen and passivate defects on the surface. The FG pretreatment represents a novel, useful, and easy method to increase the passivation quality of the c-Si wafers by ALD- ${\rm TiO_X}$ with application potential in both perovskite/Si tandems and single-junction Si solar cells.

ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge at https://pubs.acs.org/doi/10.1021/acsaem.0c00378.

Experimental details, additional figures including ${\rm TiO_2}$ thickness effect, stability test, PL spectra, various preheat treatment conditions effects, passivation results for samples without boron diffusion and with phosphorus diffusion, elemental map, simulation result, energy level structure, and tables including Si species' relative intensities and ${\rm Ti^{3^+}}$ defect rate (PDF)

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Notes

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